

1. Consider the following differentials:

$$\dot{d}u = -\frac{y}{x^2 + y^2}dx + \frac{x}{x^2 + y^2}dy$$

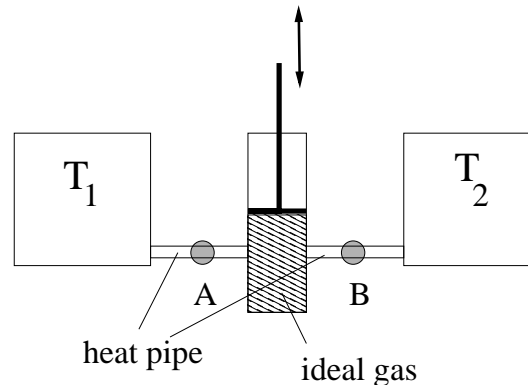
$$\dot{d}u = \frac{y}{z}dx + \frac{x}{z}dy - \frac{xy}{z^2}dz$$

$$\dot{d}u = (2y^2 - 3x)dx - 4xy dy$$

$$\dot{d}u = (x + y/x)dx + dy$$

Which of these are exact, and in case of inexact differentials, find the integrating factor.

2. Calculate the Legendre transformation from variable  $x$  to  $y$ ;  $f(x) \rightarrow g(y) = f(x(y)) - xy$ ,  $y = \partial f / \partial x$ , when
- $f(x) = e^x$
  - $f(x) = x^n$  for integer  $n$
3. Calculate the virial expansion coefficients for the Van der Waals equation of state. Show that the second coefficient is  $B_2(T) = b - \frac{a}{k_B T}$ .
4. A realization of the Carnot engine: consider a piston filled with ideal gas connected to 2 heat reservoirs at temperatures  $T_1$  and  $T_2$ , with  $T_2 > T_1$ . The connection is via “heat pipes”, which are equipped with switches A and B. When both switches are disconnected the piston is thermally isolated.



What are the 4 stages of the Carnot cycle in this setup? Let us assume that during the isothermal stage at  $T = T_2$  (stage  $a$  in the lecture notes) the volume of the system expands from  $V_1 \rightarrow V_2$ . Calculate  $\Delta W$ ,  $\Delta Q$  and  $\Delta U$  during the 4 stages of the cycle.

You will need the ideal gas equation of state,  $pV = Nk_B T$ , and the expression for the internal energy  $U = \frac{3}{2}Nk_B T$  (for monatomic ideal gas).