- 1. Write the van der Waals equation of state in terms of molecular volume  $v =$  $V/N$  (inverse of density). Sketch the behaviour of isotherms  $p(v, T = \text{const.})$ at large and small T. What is the critical point  $(T_c, p_c, v_c)$ ?
- 2. a) Calculate the phase diagram of van der Waals gas on  $(T, p)$ -plane in close proximity of the critical point (i.e. deviations from the critical quantitites are small).
	- b) Find the discontinuity of the density  $n = N/V$  near the critical point, i.e. find  $\Delta n = n_{+}(p,T) - n_{-}(p,T)$ , where  $n_{\pm}$  are the densities of liquid and gas along the coexistence line. If we write  $\Delta n \propto (T_c - T)^{\beta}$ , find the value of the exponent  $\beta$ .
	- c) Find the latent heat  $\Delta H$  near the critical point.
- 3. Show that the ensemble probability density function  $\rho$  obeys the Liouville equation

$$
i\frac{\partial \rho}{\partial t} = L\rho \,,
$$

where  $L$  is the Liouville operator

$$
L = i\{H, \quad\} = i\sum_{j} \left( \frac{\partial H}{\partial q_{j}} \frac{\partial}{\partial p_{j}} - \frac{\partial H}{\partial p_{j}} \frac{\partial}{\partial q_{j}} \right).
$$

If  $H$  is independent of  $t$ , show that the formal solution of the equation is

$$
\rho(t) = e^{-iLt}\rho(0).
$$

4. Let us assume that the probability distribution depends on P only through Hamilton function  $H$ ,

$$
\rho(P) = \rho(H[P]).
$$

Show that the distribution is stationary,  $\partial \rho / \partial t = 0$ , and that the expectation value of an observable f with no explicit time dependence is also stationary:

$$
\frac{d\langle f \rangle}{dt} = 0.
$$